Hydrothermal Synthesis of a Novel Thermally Stable Three-Dimensional Ytterbium-Organic Framework

Supporting Information

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Fig. S1 Scanning Electron Microscopy (SEM) images of CUmof-9 obtained using a FEG-SEM Hitachi S4100 microscope operating at 25 kV. Samples were prepared by depositing CUmof-9 on aluminium foil and carbon coating.



Fig. S2 Comparison of the experimental (top) and simulated (bottom) powder X-ray patterns for CUmof-9. Data were collected at ambient temperature using the step counting method (step 0.5° , time 460s) on a STOE STADI-P high-resolution transmission diffractometer with Ge(111)-monochromated Cu K α radiation (λ =1.5406 Å), and a position-sensitive detector covering a 6° 2 θ angle (40 kV, 40 mA). The simulated powder pattern was based on single-crystal data and calculated using the STOE Win XPOW software package.



Fig. S3 Thermal analysis of CUmof-9: TG, DTG and DSC curves. The compound loses almost all the crystallisation water below *ca.* 200°C, with the oxidation of the framework starting only above *ca.* 550°C. TGA was carried out using a Shimadzu TGA-50 and DSC on a Shimadzu DSC-50 analysers. Measurements were performed at a heating rate of 5°C/min under a nitrogen atmosphere with a flow rate of 20 cm³/min.





Absorption of water by the dehydrated CUmof-9 was performed under a humid atmosphere inside a desiccator containing a saturated solution of NH_4NO_3 . Elemental analysis: C 41.96%, H 2.27%.



Fig. S5 Comparison of the FT-IR spectra for dehydrated (bottom) and re-hydrated (top) CUmof-9. FT-IR spectra were collected using a Thermo Nicolet spectrometer (NEXUS family) equipped with a Smart Golden Gate for ATR analyses.